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(54) **Inorganic cohesion agent for self-compacting cement pastes**

(57) The present invention relates to an inorganic cohesion agent for self-compacting cement pastes consisting of a co-precipitated  $\text{SiO}_2/\text{CaCO}_3$  mixture.

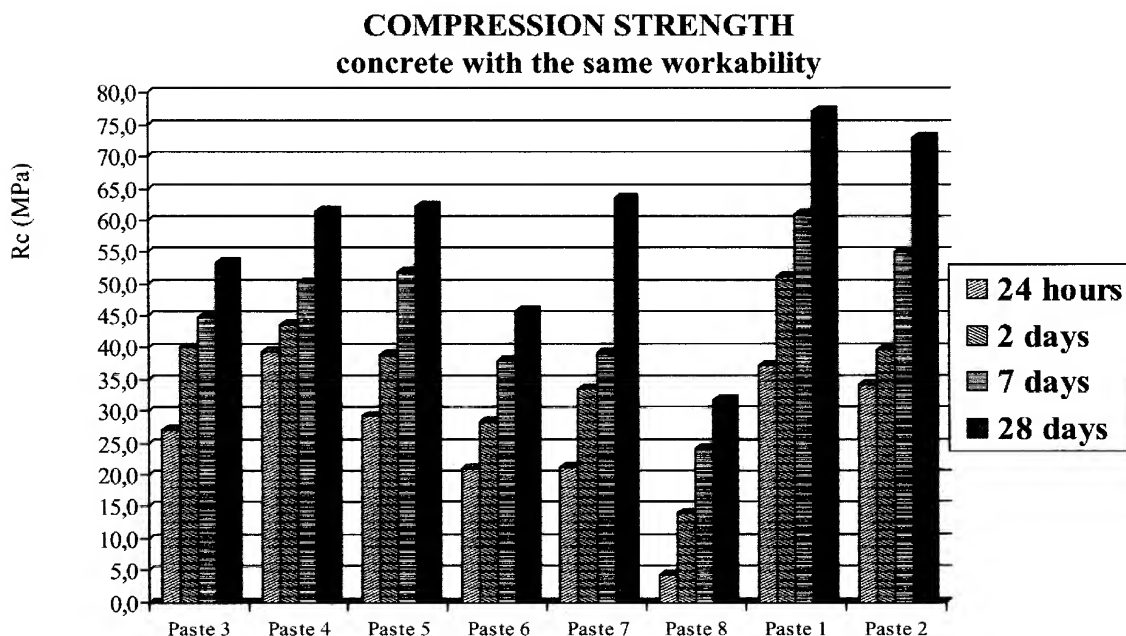


fig. 9

## Description

[0001] The present invention relates to an inorganic cohesion agent for self-compacting cement pastes.

[0002] Self-compacting concrete (SCC), which was developed in Japan in the eighties', is well known in literature (Okamura, H.; Kunishima, M.; Maekawa, K. and Ozawa, K.; High Performance Concrete based on the Durability Design of Concrete Structures, Proceedings of EASEC-2, Nr. 1, January 1989, pages 445-450). This self-compacting concrete consists of cement conglomerates whose fluidity (self-levelling), in the fresh state, is such that they can be used without any need for vibration or compacting stress.

[0003] These self-compacting cement pastes must have excellent properties both when fresh (fluidity, cohesion and absence of segregation), and in the hardened state (mechanical resistance and durability). These characteristics can be obtained by contemporaneously adopting a low water/cement ratio, the use of super-fluidifying additives and the addition of cohesion or viscosizing agents.

[0004] The main characteristic of self-compacting concrete, as it is self-levelling, is an extremely high fluidity: this means that the slump test measurement is so high (>240 mm) as to be no longer significant; resort is therefore made to the slump flow measurement which must reach values of at least 600 mm.

[0005] The other essential characteristic of a self-compacting concrete, in addition to a considerably high fluidity, is the absence of bleeding and segregation. The absence of segregation is obtained by the addition of extremely effective mineral fillers such as, for example, silica fume and/or viscosizing agents of an organic nature, such as products based on modified cellulose.

[0006] As far as the properties in the hardened state are concerned, adequate mechanical resistance, in particular compression strength, and good durability of concrete structures can be obtained in a material with a high compacting degree.

[0007] The durability of concrete is correlated to the permeability in relation to the homogeneity of the phases of which it is composed (matrix, aggregate, transition zone). A reduced permeability of the material together with uniformity of the permeability values in the different zones of a cementitious end-product, indicate a good compacting degree.

[0008] There is no absolute measurement of a permeability coefficient, but there are various methods which measure different fluid transport mechanisms through the material. One of these methods is gas permeability, in particular air permeability.

[0009] The objective of the present invention is to identify an alternative agent to those existing in the state of the art for providing cohesion and compacting characteristics, thus obtaining a self-compacting concrete which contemporaneously has a high stability, viscosity and rest cohesion (i.e. absence of bleeding and segregation) and good compression strength values.

[0010] An object of the present invention relates to an inorganic cohesion agent for self-compacting cement pastes, consisting of a co-precipitated  $\text{SiO}_2/\text{CaCO}_3$  mixture.

[0011] In particular, the composite material made up of the co-precipitated mixture of silica and calcium carbonates which forms the agent according to the present invention is obtained starting from natural or synthetic calcium silicates, crystalline or amorphous, hydrated or non-hydrated, or their mixtures, which, by reaction with  $\text{CO}_2$  in water, can allow solid co-precipitated mixtures of  $\text{SiO}_2$  and  $\text{CaCO}_3$  to be obtained.

[0012] The starting materials can also be cements or cement clinkers.

[0013] The co-precipitated  $\text{SiO}_2/\text{CaCO}_3$  mixtures according to the present invention can be used as such, as deriving from the production process, or in aqueous suspension, or they can be used in the dry state, after evaporation/separation of the whole or part of the water.

[0014] The co-precipitated  $\text{SiO}_2/\text{CaCO}_3$  mixture according to the present invention preferably has a weight ratio  $\text{SiO}_2/\text{CaCO}_3$  ranging from 1.2 to 0.1. It can contain up to 50% of other components in mass, among which calcium silicates (non-reacted or transformed), for example those indicated as CSH, and/or metallic oxides.

[0015] The specific surface measured with the BET method can range from 20 to 100  $\text{m}^2/\text{g}$ .

[0016] An example of a process for the production of silica and silica and calcium carbonate composites, starting from calcium silicates is described in the patent CA 1,122,779. This patent describes a process for the production of silica, in which calcium silicate crystals are put in contact with  $\text{CO}_2$  in the presence of water and converted into silica, having the same configuration as silicate crystals, and into calcium carbonate particles attached to amorphous silica particles.

[0017] This  $\text{SiO}_2/\text{CaCO}_3$  mixture (i.e. the mixture according to the patent) is preferably treated with inorganic acids in order to decompose the calcium carbonate, separate the calcium salts and obtain pure amorphous silica or it is used as such without any type of treatment or separation.

[0018] A further object of the present invention is a self-compacting concrete obtained with the use of an inorganic cohesion agent consisting of a co-precipitated  $\text{SiO}_2/\text{CaCO}_3$  mixture.

[0019] The present invention also relates to recovery mortars and pastes which can be obtained by the use of an inorganic cohesion agent consisting of a co-precipitated  $\text{SiO}_2/\text{CaCO}_3$  mixture.

[0020] The co-precipitated  $\text{SiO}_2/\text{CaCO}_3$  mixture is preferably used in doses ranging from 1 to 30% with respect to the weight of the cement, in particular from 5 to 15%.

[0021] The main advantage of the agent according to the present invention lies in its being a cohesion agent which, as well as guaranteeing segregation-absence properties in the concrete better than or equivalent to those provided by the additives and/or mineral fillers normally used, at the same time allows the production of a long-lasting concrete with excellent compression strength.

[0022] In fact, although the use of some additives, such as precipitated silica alone or an organic viscosizing agent, on the one hand allows better cohesion of pastes in the fresh state, on the other hand it causes a drastic reduction in the compression strength with respect to a concrete with the same workability (free spreading), prepared with cement alone and without additives. Although concrete prepared with other mineral fillers such as calcareous fillers, has sufficiently high compression strength values ( $R_c$ ), there are segregation phenomena in the fresh state. This segregation causes a lack of homogeneity in an end-product both during its laying and after hardening, with the formation of zones having different compacting degrees, thus showing different permeability values, in particular different air permeability coefficients.

[0023] The characteristics and advantages of the agent according to the present invention are better illustrated by the following detailed description, referring to the following examples.

[0024] Description of the enclosed figures:

Figures 1-4 are a photographic representation of pastes 2, 3, 4 and 9 respectively, in the fresh state;  
Figures 5, 6 and 7 are a photographic representation of pastes 1, 3 and 4 respectively, in the hardened state;  
Figure 8 is a photographic representation which shows in order test samples of pastes 1, 6, 4 and 3, after ejection;  
Figure 9 is a graph which represents a comparison between the compression strengths of pastes 1-8.

[0025] Nine pastes were in fact prepared, some of which containing the agent according to the present invention (pastes 1 and 2) and others containing additives according to the state of the art (pastes 3-9).

[0026] All the pastes were prepared with equal workability (slump flow ranging from 600 to 650 mm).

[0027] In particular, the materials with which the various pastes were prepared are the following:

cement: CEM I 52, 5 R ULTRACEM Italcementi;

aggregate: silico-calcareous SATAF, subdivided into five distinct sizes. The granulometric curve is of the discontinuous type with a maximum diameter equal to 20 mm;

Sand	% with respect to the total	Particle size (mm)
113	10	0.1 ÷ 0.5
103	10	0.5 ÷ 2
109	20	8 ÷ 15
10-15	30	10 ÷ 15
15-20	40	15 ÷ 20

acrylic superfluidifying additive: 2000AC AXIM.

[0028] The concrete is prepared using a forced mixer with a vertical axis and the characteristics were evaluated by means of the following methods:

volume mass ( $\text{kg/m}^3$ ): UNI 6394 method, 1<sup>st</sup> part;

free spreading (cm): slump test;

compression strength (MPa): EN 196/1 method;

air permeability: on samples having  $\Phi = 8$  cm and  $h = 3$  cm, taken from the upper part and from the lower part of cylindrical test samples ( $\Phi = 12$  cm and  $h = 40$  cm), after 28 days of curing. The cylindrical test samples were obtained by filling appropriate cylindrical moulds with concrete, without vibration, and leaving it to harden in a vertical position. For the air permeability measurements, a Hassler-type cell was used, calculating the permeability coefficient  $k$  by means of the following equation (proposed by Grube and Lawrence):

$$k = \frac{2\eta \cdot L \cdot V_2 \cdot P_2}{(p_1^2 - p_2^2) \cdot A};$$

wherein

k = permeability coefficient, m<sup>2</sup>;  
 η = viscosity of the fluid medium, Nsm<sup>-2</sup>;  
 V<sub>2</sub> = flow rate at the outlet, ms<sup>-1</sup>;  
 P<sub>1</sub> and P<sub>2</sub> = inlet and outlet pressure respectively, Nm<sup>-2</sup>;  
 A = transversal surface of the test sample, m<sup>2</sup>;  
 L = thickness of the sample, m;

cohesion and segregation evaluation: this evaluation was effected both in the fresh state, by evaluating the cohesion of the paste after its preparation and the possible presence of bleeding, and also in the hardened state, by evaluating the internal segregation of cylindrical test samples (Φ = 12 cm; h = 40÷50 cm), specifically prepared without vibration and broken by indirect tensile stress after two days of hardening;  
 aesthetic evaluation: visual evaluation of the cylindrical test-samples, after hardening.

**[0029]** The cohesion agents present in pastes 1-9 of the following examples, are:

Paste 1: cohesion agent according to the present invention, co-precipitated SiO<sub>2</sub>/CaCO<sub>3</sub> mixture alone;  
 Paste 2: cohesion agent according to the present invention, SiO<sub>2</sub>/CaCO<sub>3</sub>/CSH mixture;  
 Paste 3 and 3 bis: no cohesion agent;  
 Paste 4: Carrara calcareous filler (comparative)  
 Paste 5: commercial precipitated CaCO<sub>3</sub> (comparative);  
 Paste 6: commercial precipitated silica ULTRASIL VN3 (Degussa) (comparative);  
 Paste 7: Elkem 940 Silica fume (comparative);  
 Paste 8: commercial organic viscosizing agent (comparative);  
 Paste 9: no cohesion agent, but increase in the cement dosage.

#### EXAMPLE 1

**[0030]** A 20 litre paste was prepared with the following composition:

PASTE 1 (with the agent according to the present invention, co-precipitated  $\text{SiO}_2/\text{CaCO}_3$  mixture alone).

Composition	Weight (kg)	Dosage (kg/m <sup>3</sup> )
aggregate	38.000	1900
additions ( $\text{SiO}_2/\text{CaCO}_3$ mixture)	0.600	30
Cement	8.000	400
paste water	3.700	185
acrylic additive	0.160	8.0
water*/cement ratio	0.48	
water*/(cement + additions) ratio	0.44	

\* the water also comprises the water contained in the acrylic additive.

[0031] Paste 1 has the following characteristics:

volume mass: 2405 kg/cm<sup>3</sup>;  
 free spreading: 63 cm;  
 compression strength: see Table 1 below;  
 segregation evaluation: see figure 5;  
 aesthetic evaluation: after ejection, the test sample has a good aesthetic appearance (figure 8).

[0032] The use of the co-precipitated  $\text{SiO}_2/\text{CaCO}_3$  mixture alone, according to the present invention, with dosages ranging from 1 to 30% with respect to the weight of the cement, in particular from 5 to 15%, allows a cohesive and non-segregable system to be obtained, of extremely easy application in the self-compacting concrete field.

TABLE 1

Paste	Rc (Mpa)			
	1 day	2 days	7 days	28 days
Paste 1	37.2	51.2	61.0	77.1
Paste 2	34.2	39.7	55.0	73.0
Paste 3	27.2	40.0	44.8	53.4

TABLE 1 (continued)

Paste	Rc (Mpa)			
	1 day	2 days	7 days	28 days
Paste 3 bis	39.8	41.8	51.1	65.0
Paste 4	39.4	43.5	50.3	61.5
Paste 5	39.2	51.0	55.0	68.9
Paste 6	32.6	38.4	49.2	55.7
Paste 7	21.3	33.6	39.2	63.5
Paste 8	4.4	14.0	24.2	31.8
Paste 9	33.4	46.7	55.5	66.2

EXAMPLE 2

[0033] A 20 litre paste was prepared with the following composition:

PASTE 2 (with the agent according to the present invention, SiO<sub>2</sub>/CaCO<sub>3</sub>/CSH mixture).

Composition	Weight (kg)	Dosage (kg/m <sup>3</sup> )
aggregate	38.000	1900
additions (SiO <sub>2</sub> /CaCO <sub>3</sub> /CSH mixture)	0.600	30
Cement	8.000	400
paste water	4.200	210
acrylic additive	0.160	8.0
water*/cement ratio	0.54	
water*/(cement + additions) ratio	0.50	

\* the water also comprises the water contained in the acrylic additive.

[0034] Paste 2 has the following characteristics:

volume mass: 2373 kg/cm<sup>3</sup>;

free spreading: 60 cm;

compression strength: see table 1;

evaluation of the air permeability coefficient  $k$  (m<sup>2</sup>) on samples having  $\Phi = 8$  cm and  $h = 3$  cm, which form the upper and lower part of cylindrical test samples having  $\Phi = 12$  cm and  $h = 40$  cm: see Table 2 below;

cohesion evaluation: see figure 1;

aesthetic evaluation: after ejection, the test sample has a good aesthetic appearance.

TABLE 2

Paste	k (x 10 <sup>-17</sup> m <sup>2</sup> )	
	upper	lower
Paste 2	1.8	1.1
Paste 3	23.1	1.3
Paste 4	12.0	1.0
Paste 5	17.4	4.7
Paste 8	30.6	6.7

### EXAMPLE 3

**[0035]** A 20 litre paste was prepared with the following composition:

#### PASTE 3 (without cohesion agent).

Composition	Weight (kg)	Dosage (kg/m <sup>3</sup> )
aggregate	38.000	1900
additions	-	-
cement	8.000	400
paste water	3.500	175
acrylic additive	0.136	6.8
water*/cement ratio	0.45	
water*/(cement + additions) ratio	0.45	

\* the water also comprises the water contained in the acrylic additive.

**[0036]** The composition of the paste was selected so as to have a paste with a high segregation in order to enhance the improvement obtained as a result of the cohesion additives.

[0037] Paste 3 has the following characteristics:

volume mass: 2443 kg/m<sup>3</sup>;  
 free spreading: 63 cm;  
 compression strength: see table 1;  
 cohesion and segregation evaluation: see figures 2 and 6; aesthetic evaluation: after ejection, the test sample has a good aesthetic appearance (figure 8).

[0038] Example 3 was subsequently repeated with a 20 litre paste having the following composition:

PASTE 3 bis (without cohesion agent).

Composition	Weight (kg)	Dosage (kg/m <sup>3</sup> )
aggregate	38.000	1900
additions	-	-
cement	8.600	430
paste water	3.800	190
acrylic additive	0.136	6.8
water*/cement ratio	0.45	
water*/(cement + additions) ratio	0.45	

[0039] Paste 3 bis has the following characteristics:

volume mass: 2429 kg/m<sup>3</sup>;  
 free spreading: 63 cm;  
 compression strength: see table 1;  
 aesthetic evaluation: after ejection, the test sample has a good aesthetic appearance.

EXAMPLE 4

[0040] A 20 litre paste was prepared with the following composition:



PASTE 4 (with Carrara calcareous filler).

Composition	Weight (kg)	Dosage (kg/m <sup>3</sup> )
aggregate	37.000	1850
additions (calcareous filler)	1.000	50
cement	8.000	400
paste water	3.500	175
acrylic additive	0.136	6.8
water*/cement ratio	0.45	
water*/(cement + additions) ratio	0.40	

\* the water also comprises the water contained in the acrylic additive.

**[0041]** Paste 4 has the following characteristics:

volume mass: 2444 kg/m<sup>3</sup>;  
 free spreading: 64 cm;  
 compression strength: see table 1;  
 cohesion and segregation evaluation: see figures 3 and 7; aesthetic evaluation: after ejection, the test sample has a good aesthetic appearance (figure 8).

**[0042]** It is evident from the photographs provided in figures 3 and 7, that a calcareous filler is not capable of increasing the cohesion of a "segregable" concrete such as that of Example 3, represented in figures 2 and 6.

EXAMPLE 5

**[0043]** A 20 litre paste was prepared with the following composition:

PASTE 5 (with precipitated  $\text{CaCO}_3$ ).

Composition	Weight (kg)	Dosage (kg/m <sup>3</sup> )
aggregate	38.000	1900
additions (precipitated $\text{CaCO}_3$ )	0.480	24
Cement	8.000	400
paste water	3.700	185
acrylic additive	0.136	6.8
water*/cement ratio	0.47	
water*/(cement + additions) ratio	0.45	

\* the water also comprises the water contained in the

acrylic additive.

**[0044]** Paste 5 has the following characteristics:

volume mass: 2339 kg/cm<sup>3</sup>;

free spreading: 63 cm;

compression strength: see table 1;

evaluation of the air permeability coefficient  $k$  (m<sup>2</sup>): see Table 2;

aesthetic evaluation: after ejection, the test sample has a good aesthetic appearance.

EXAMPLE 6

**[0045]** A 20 litre paste was prepared with the following composition:

PASTE 6 (with silica VN<sub>3</sub>).

Composition	Weight (kg)	Dosage (kg/m <sup>3</sup> )
aggregate	38.000	1900
additions (silica VN <sub>3</sub> )	0.240	12
Cement	8.000	400
paste water	4.350	217.5
acrylic additive	0.160	8.0
water*/cement ratio	0.56	
water*/(cement + additions) ratio	0.54	

\* the water also comprises the water contained in the acrylic additive.

**[0046]** Paste 6 has the following characteristics:

volume mass: 2397 kg/cm<sup>3</sup>;  
 free spreading: 60 cm;  
 compression strength: see table 1;  
 aesthetic evaluation: after ejection, the test sample has a good aesthetic appearance (figure 8).

EXAMPLE 7

**[0047]** A 20 litre paste was prepared with the following composition:

PASTE 7 (with silica fume).

Composition	Weight (kg)	Dosage (kg/m <sup>3</sup> )
aggregate	38.000	1900
additions (silica fume)	0.900	45
Cement	8.000	400
paste water	4.600	230
acrylic additive	0.160	8.0
water*/cement ratio	0.59	
water*/(cement + additions) ratio	0.53	

\* the water also comprises the water contained in the acrylic additive.

[0048] Paste 7 has the following characteristics:

volume mass: 2408 kg/cm<sup>3</sup>;  
 free spreading: 60 cm;  
 compression strength: see table 1;  
 aesthetic evaluation: after ejection, the test sample has a good aesthetic appearance.

EXAMPLE 8

[0049] A 20 litre paste was prepared with the following composition:

PASTE 8 (with organic viscosizing agent).

Composition	Weight (kg)	Dosage (kg/m <sup>3</sup> )
aggregate	38.000	1900
additions (organic viscosizing agent Kelko-Krete)	0.008	0.4
Cement	8.000	400
paste water	5.400	270
acrylic additive	0.160	8.0
water*/cement ratio	0.69	
water*/(cement + additions) ratio	0.69	

\* the water also comprises the water contained in the acrylic additive.

**[0050]** Paste 8 has the following characteristics:

volume mass: 2365 kg/cm<sup>3</sup>;  
 free spreading: 60 cm;  
 compression strength: see table 1;  
 evaluation of the air permeability coefficient k (m<sup>2</sup>): see Table 2.

EXAMPLE 9

**[0051]** A 20 litre paste was prepared with the following composition:

PASTE 9 (without a cohesion agent, but with an increase  
in the dosage of cement).

Composition	Weight (kg)	Dosage (kg/m <sup>3</sup> )
aggregate	36.000	1800
additions	-	-
Cement	10.000	500
paste water	4.000	200
acrylic additive	0.170	8.5
water*/cement ratio	0.41	
water*/(cement + additions) ratio	0.41	

\* the water also comprises the water contained in the acrylic additive.

[0052] Paste 9 has the following characteristics:

volume mass: 2425 kg/cm<sup>3</sup>;  
free spreading: 64 cm;  
compression strength: see table 1;  
cohesion evaluation: see figure 4;  
aesthetic evaluation: after ejection, the test sample has a good aesthetic appearance.

[0053] From a comparison of the previous data, the following can be observed.

[0054] Pastes 1 and 6, i.e. the pastes containing, as cohesion agent, the agent according to the present invention and commercial precipitated silica, do not undergo any decomposition in the absence of vibration, which on the contrary occurs with pastes 3, 4 and 9, as can be seen from figures 1-4 and 5-7.

[0055] With the same workability with respect to pastes 1, 3 and 4, however, paste 6, i.e. with commercial precipitated silica, requires an increase in the paste water in the order of 25% and in the acrylic superfluidifying agent of 18%.

[0056] The use, on the other hand, of the co-precipitated SiO<sub>2</sub>/CaCO<sub>3</sub> mixture, according to the present invention, limits the increase in the quantity of paste water to 13%. The influence in the water/cement ratio negatively influences the compression strength of concrete with commercial silica, i.e. paste 6. This effect, as can be seen from the graph in figure 9, is not observed for paste 1 which, for all the tests, has values similar to those of pastes 3 and 4.

[0057] This result is probably due to a rapid activation of the silica in the co-precipitated SiO<sub>2</sub>/CaCO<sub>3</sub> mixture, together with a "filler" effect of the co-precipitated SiO<sub>2</sub>/CaCO<sub>3</sub> mixture.

[0058] Figure 8 shows that pastes 1 and 6 have the best finishing.

[0059] On comparing the compression strength values of the various pastes indicated in Table 1, it can be immediately seen how, with the same workability, pastes 1 and 2, i.e. the pastes containing the cohesion agent according to the

present invention, have a very high compression strength with respect to the pastes with other additives according to the state of the art.

[0060] From the data provided in Table 2, it can be observed that paste 2, containing the cohesion agent according to the present invention, has very low air permeability coefficient values and, above all, practically constant values at all points of the test sample. The pastes containing additives according to the state of the art, on the contrary, have extremely varying permeability coefficient values between different points of the test sample and this lack of homogeneity is a further confirmation of the fact that they do not have a good compacting degree.

[0061] The main advantage of the cohesion agent according to the present invention is that, in addition to guaranteeing non-segregation properties in the concrete which are better than or equivalent to those provided by the additives normally used, it also allows a long-lasting concrete to be obtained, with an excellent compression strength.

## Claims

1. An inorganic cohesion agent for self-compacting cement pastes, consisting of a co-precipitated  $\text{SiO}_2/\text{CaCO}_3$  mixture.
2. The cohesion agent according to claim 1, **characterized in that** the co-precipitated mixture of silica and calcium carbonates is obtained starting from natural or synthetic calcium silicates, crystalline or amorphous, hydrated or non-hydrated, or their mixtures, cements or cement clinkers.
3. The cohesion agent according to claim 1, **characterized in that** the co-precipitated  $\text{SiO}_2/\text{CaCO}_3$  mixture is in aqueous suspension.
4. The cohesion agent according to claim 1, **characterized in that** the co-precipitated  $\text{SiO}_2/\text{CaCO}_3$  mixture is in the dry state.
5. The cohesion agent according to claim 1, **characterized in that** the co-precipitated  $\text{SiO}_2/\text{CaCO}_3$  mixture has a weight ratio  $\text{SiO}_2/\text{CaCO}_3$  ranging from 1.2 to 0.1.
6. The cohesion agent according to claim 1, **characterized in that** the co-precipitated  $\text{SiO}_2/\text{CaCO}_3$  mixture contains up to 50% of other components in mass.
7. The cohesion agent according to claim 6, **characterized in that** the other components are calcium silicates (non-reacted or transformed), for example those indicated as CSH, and/or metallic oxides.
8. The cohesion agent according to claim 1, **characterized in that** the specific surface of the co-precipitated mixture ranges from 20 to 100  $\text{m}^2/\text{g}$ .
9. The cohesion agent according to claim 1, **characterized in that** the co-precipitated  $\text{SiO}_2/\text{CaCO}_3$  mixture is present in dosages ranging from 1 to 30% with respect to the weight of the cement.
10. The cohesion agent according to claim 1, **characterized in that** the co-precipitated  $\text{SiO}_2/\text{CaCO}_3$  mixture is present in dosages ranging from 5 to 15% with respect to the weight of the cement.
11. The cohesion agent according to claim 1, which can be obtained by means of a process for the production of silica, in which calcium silicate crystals are put in contact with  $\text{CO}_2$  in the presence of water and converted to silica, having the same configuration as silicate crystals, and to calcium carbonate particles attached to amorphous silica particles.
12. Use of the agent according to any of the previous claims, for self-compacting cement pastes.
13. Self-compacting concrete obtained with the use of an inorganic cohesion agent according to any of the claims from 1 to 10.
14. Recovery mortars or pastes obtained by the use of an inorganic cohesion agent according to any of the claims from 1 to 10.

**COHESION EVALUATION IN THE FRESH STATE**  
**Paste 2: concrete with experimental  $\text{SiO}_2/\text{CaCO}_3/\text{CSH}$  mixture**



fig. 1



# COHESION EVALUATION IN THE FRESH STATE

Paste 3: concrete as such



fig. 2

## **COHESION EVALUATION IN THE FRESH STATE**

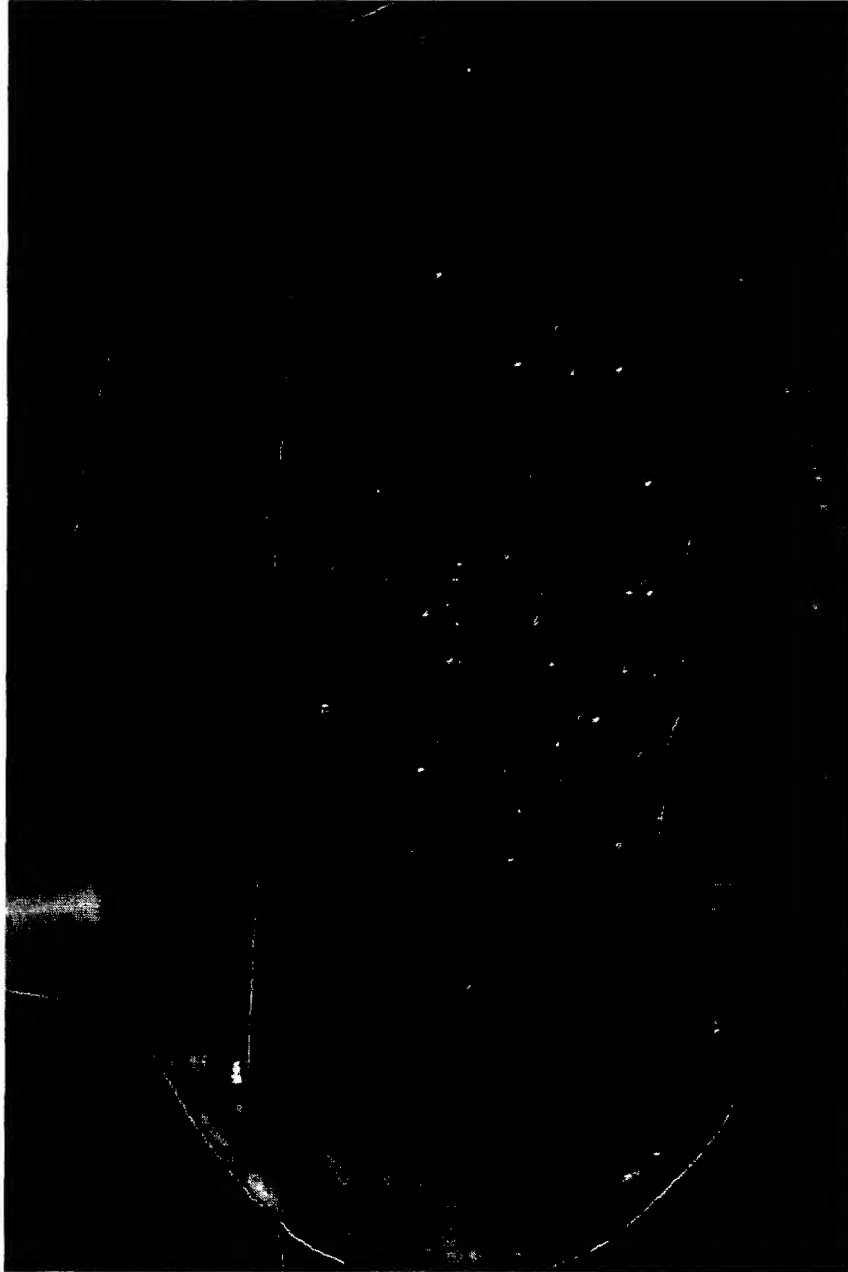
**Paste 4: concrete with calcareous filler**



**fig. 3**

## **COHESION EVALUATION IN THE FRESH STATE**

**Paste 9: concrete as such - increase in cement dosage**



**fig. 4**

## SEGREGATION EVALUATION

Concrete in the hardened state

TOP

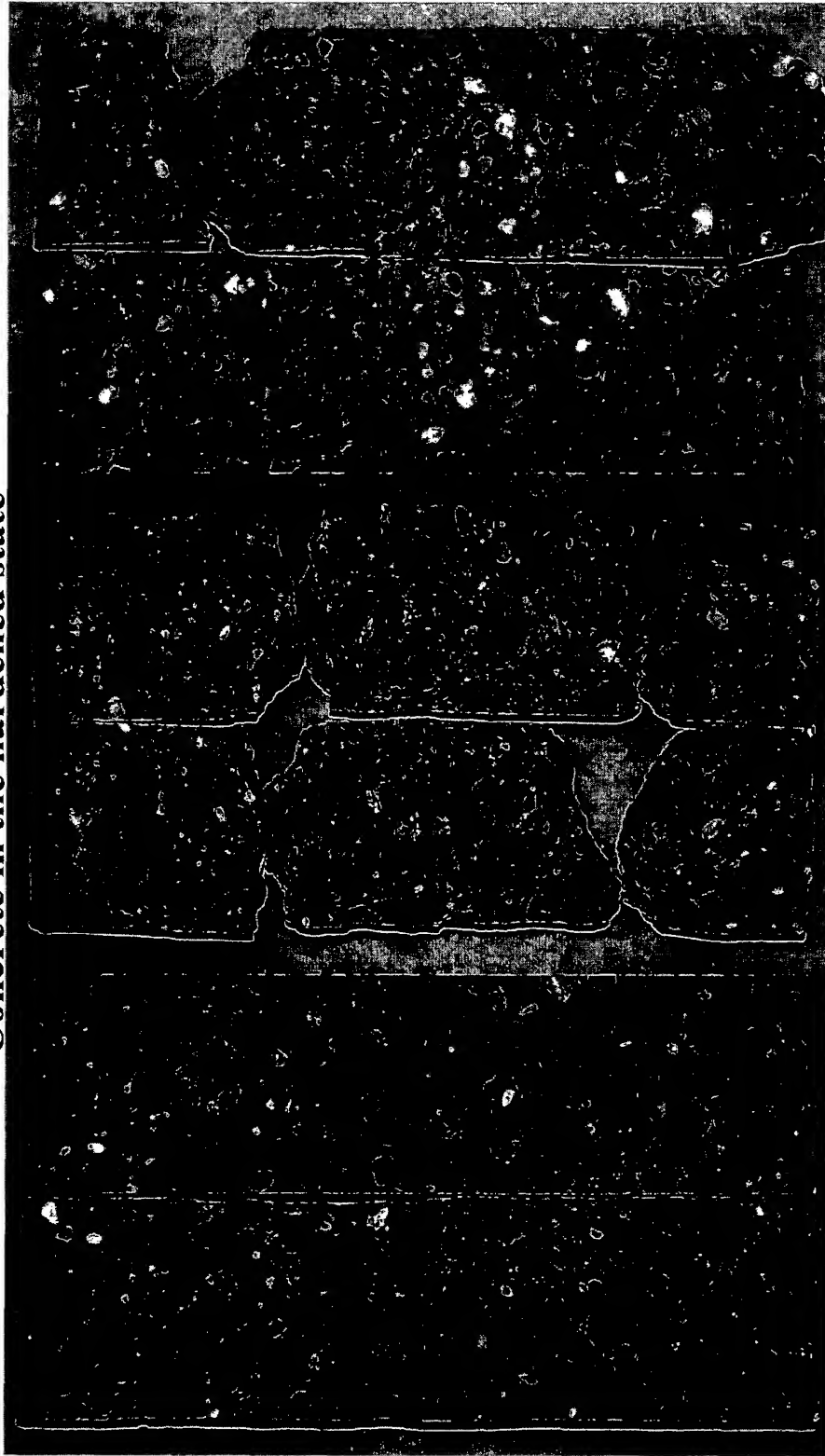


Figure 5: paste 1-with the mixture  
according to the invention

Figure 6: paste 3 -as such

Figure 7: paste 4-with calcareous filler

BOTTOM

**AESTHETIC EVALUATION  
concrete in the hardened state**



**fig. 8**

# COMPRESSION STRENGTH concrete with the same workability

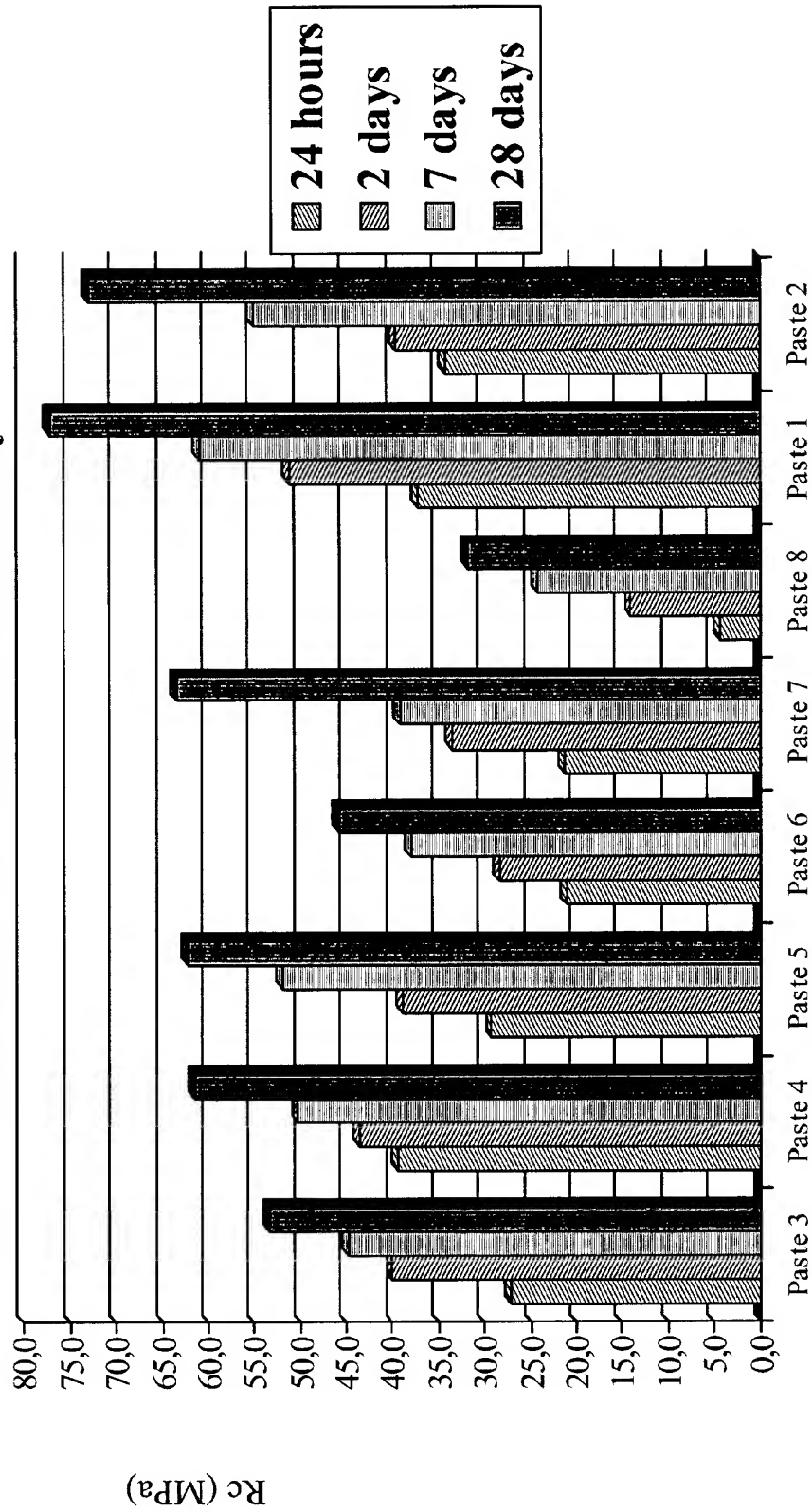


fig. 9



European Patent  
Office

## EUROPEAN SEARCH REPORT

Application Number  
EP 01 20 2740

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
P,X	WO 00 46149 A (GRONCHI PAOLO ;CASSAR LUIGI (IT); ITALCEMENTI SPA (IT); MARCO TIZI) 10 August 2000 (2000-08-10) * page 8, line 14 - page 9, line 4 * * page 10, line 4 - line 24 * ---	1-11	C04B14/28 C04B14/06 C04B28/02 C01B33/193 C01B11/18
A	EP 0 744 387 A (MARUTAKA CONCRETE INDUSTRY CO ;NISSAN CHEMICAL IND LTD (JP); TAKAG) 27 November 1996 (1996-11-27) * page 2, line 55 - page 3, line 7 * ---	1,12,13	
A	DATABASE WPI Section Ch, Week 197652 Derwent Publications Ltd., London, GB; Class E33, AN 1976-96970X XP002182095 & JP 51 129424 A (OSAKA PACK MFG KK), 11 November 1976 (1976-11-11) * abstract * ---	1,11	
A	EP 0 323 577 A (KNAUF WESTDEUTSCHE GIPS) 12 July 1989 (1989-07-12) * page 2, right-hand column, line 50 - page 3, left-hand column, line 5 * ---	1	TECHNICAL FIELDS SEARCHED (Int.Cl.7)  C04B C01B
A	PATENT ABSTRACTS OF JAPAN vol. 1999, no. 08, 30 June 1999 (1999-06-30) & JP 11 079818 A (DENKI KAGAKU KOGYO KK), 23 March 1999 (1999-03-23) * abstract * ---	1	
A	FR 2 746 096 A (RHONE POULENC CHIMIE) 19 September 1997 (1997-09-19) * page 4, line 1 - line 23 * -----	1	
The present search report has been drawn up for all claims			
Place of search <b>THE HAGUE</b>		Date of completion of the search <b>6 November 2001</b>	Examiner <b>Theodoridou, E</b>
CATEGORY OF CITED DOCUMENTS X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons ----- & : member of the same patent family, corresponding document	

EPO FORM 1503 (03.92) (P4/C01)

**ANNEX TO THE EUROPEAN SEARCH REPORT  
ON EUROPEAN PATENT APPLICATION NO.**

EP 01 20 2740

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on  
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

06-11-2001

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 0046149 A	10-08-2000	AU 2800100 A	25-08-2000
		WO 0046149 A1	10-08-2000
		EP 1149047 A1	31-10-2001
EP 0744387 A	27-11-1996	JP 8325049 A	10-12-1996
		AU 699201 B2	26-11-1998
		AU 5207696 A	05-12-1996
		CA 2176410 A1	27-11-1996
		DE 69601896 D1	06-05-1999
		DE 69601896 T2	21-10-1999
		EP 0744387 A1	27-11-1996
		US 5676749 A	14-10-1997
JP 51129424 A	11-11-1976	JP 1184225 C	27-12-1983
		JP 55023790 B	25-06-1980
EP 0323577 A	12-07-1989	DE 3743855 C1	03-05-1989
		AT 56692 T	15-10-1990
		EP 0323577 A1	12-07-1989
JP 11079818 A	23-03-1999	NONE	
FR 2746096 A	19-09-1997	FR 2746096 A1	19-09-1997

EPO FORM P0458

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82